

7.3. Technologies for Hydrogen Production

Technologies for hydrogen (H₂) production fall into three main categories:

1. **Thermal Processes:** Some thermal processes use the energy in various feedstocks (natural gas, coal, biomass, etc.) to release the H₂ that is part of their molecular structure. Other thermal processes known as thermo-chemical processes use heat in combination with a closed chemical cycle to produce H₂ from feedstocks such as water.

In addition to gasification, the main thermal process technology which is available for production of H₂ is steam reformation of natural gas. It is a well established technology that produces about 95% of the H₂ produced in the United States. Steam reforming involves the reaction of natural gas and steam over a nickel based catalyst. This breaks the methane component of the natural gas into carbon monoxide (CO) and H₂ gas, similar to synthesis gas (syngas) produced via gasification. Then [water-gas shift](#) (WGS) is performed to increase the amount of H₂ in the product gas as much as possible.

2. **Electrolytic Processes:** These processes use electricity to split water into its two chemical constituents, oxygen (O₂) and H₂, using an electrolyzer. The cost and efficiency of producing H₂ via electrolytic processes is directly dependent on the cost and efficiency of the electricity used in the process.
3. **Photolytic Processes:** These processes use light energy to also split water into H₂ and O₂. These processes are currently in the early stages of development and currently are not viable for large scale production.

Efficiency/Cost Comparison to Competing Technologies

Table 1 presents the cost and performance characteristics of various H₂ production pathways, as of 2004. Many of the technologies that are in the research and development (R&D) stage will require years of improvements before becoming a commercial reality.

Table 1: Efficiency/Cost Comparison to Competing Technologies ¹

Process	Energy Required (kWh/Nm ³)		Status of Tech.	Efficiency [%]	Costs Relative to SMR
	Ideal	Practical			
Steam methane reforming (SMR)	0.78	2-2.5	mature	70-80	1
Coal gasification (GE Energy)	1.01	8.6	mature	60	1.4-2.6
Partial oxidation of coal			mature	55	
H ₂ S methane reforming	1.5		R&D	50	<1
Landfill gas dry reformation			R&D	47-58	~1
Partial oxidation of heavy oil	0.94	4.9	mature	70	1.8
Naphtha reforming			mature		
Steam reforming of waste oil			R&D	75	<1
Steam-iron process			R&D	46	1.9
Chloralkali electrolysis			mature		by-product
Grid electrolysis of water	3.54	4.9	R&D	27	3-10
Solar & PV-electrolysis of water			R&D to mature	10	>3
High-temp. electrolysis of water			R&D	48	2.2
Thermochemical water splitting			early R&D	35-45	6
Biomass gasification			R&D	45-50	2.0-2.4
Photobiological			early R&D	<1	
Photolysis of water			early R&D	<10	

Photoelectrochemical decomp. of water			early R&D		
Photocatalytic decomp. of water			early R&D		

The cost of H₂ production depends heavily on the cost of fuel or electricity from which it is produced. As the market price for these inputs to the H₂ production system fluctuate, one given technology may become more attractive economically compared to others.

Coal-to-Hydrogen Process Description

The U.S. Department of Energy (DOE) has sponsored many design studies on the production of H₂ from coal, with or without the co-production of power.

Recent DOE studies presented the following four process design schemes as possible options for centralized-large-scale H₂ production from coal, and discussed their performance and efficiency:²

- Co-producing H₂ and power in today's coal-based [integrated gasification combined cycle \(IGCC\)](#) plants
- Co-producing H₂ and power in coal-based IGCC with carbon capture
- H₂ production from coal without power export
- Co-producing H₂ and power in future IGCC based on [advanced warm gas clean-up](#) and advanced membrane (combined shift and H₂ separation) technologies

1.	Hydrogen: Automotive Fuel of the Future, by FSEC's Ali T-Raissi and David Block, IEEE Power & Energy, Vol. 2, No. 6, page 43, Nov-Dec 2004.
2.	Hydrogen from Coal , D. Gray & G. Tomlinson, Mitretek Technical Paper (Nov 2001)

Hydrogen

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- **Technologies for Hydrogen Production**
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